

2-[1-(3-{2-[(2-Hydroxybenzylidene)-amino]phenoxy}propyl)-1H-1,3-benzodiazol-2-yl]phenol

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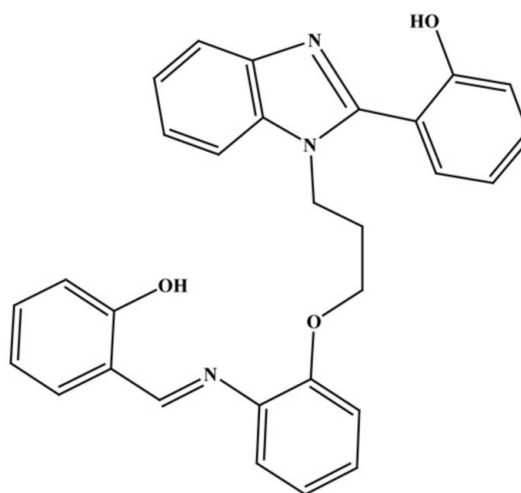
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Key indicators: single-crystal X-ray study; $T = 150$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.055; wR factor = 0.137; data-to-parameter ratio = 16.1.

In the title compound, $\text{C}_{29}\text{H}_{25}\text{N}_3\text{O}_3$, the imine double bond has an *E* configuration. The dihedral angle between the hydroxyphenyl and benzene rings in the imine moiety is $26.95(9)^\circ$, and the dihedral angle between the hydroxyphenyl and benzimidazole rings in the other moiety is $14.83(9)^\circ$. These angles are probably limited to small values as a consequence of two strong intramolecular $\text{O}-\text{H}\cdots\text{N}$ hydrogen bonds formed between the hydroxy groups and the imine and imidazole N atoms. The aliphatic chain linking the two ring systems has a *gauche* conformation, as reflected in $\text{C}-\text{C}-\text{C}-\text{O}$ torsion angle of $70.9(2)^\circ$.

Related literature

For related structures, see: Keypour *et al.* (2009). For background information on diimine complexes, see: Mahmoudi *et al.* (2009).



Experimental

Crystal data

$\text{C}_{29}\text{H}_{25}\text{N}_3\text{O}_3$

$M_r = 463.52$

Monoclinic, $P2_1/c$

$a = 9.1097(6)$ Å

$b = 18.1946(11)$ Å

$c = 13.7769(5)$ Å

$\beta = 93.405(4)^\circ$

$V = 2279.5(2)$ Å³

$Z = 4$

Mo $K\alpha$ radiation

$\mu = 0.09$ mm⁻¹

$T = 150$ K

$0.25 \times 0.12 \times 0.10$ mm

Data collection

Nonius KappaCCD diffractometer

Absorption correction: multi-scan

(*SORTAV*; Blessing, 1995)

$T_{\min} = 0.866$, $T_{\max} = 0.993$

16359 measured reflections

5135 independent reflections

3083 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.054$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.055$

$wR(F^2) = 0.137$

$S = 1.06$

5135 reflections

318 parameters

1 restraint

H-atom parameters constrained

$\Delta\rho_{\text{max}} = 0.23$ e Å⁻³

$\Delta\rho_{\text{min}} = -0.20$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{O1}-\text{H1O}\cdots\text{N2}$	0.84	1.81	2.564 (2)	148
$\text{O3}-\text{H2O}\cdots\text{N3}$	0.84	1.80	2.548 (2)	148

Data collection: *COLLECT* (Nonius, 2002); cell refinement: *DENZO-SMN* (Otwinowski & Minor, 1997); data reduction: *DENZO-SMN*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXTL* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BH2338).

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supplementary materials

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2-[1-(3-{2-[(2-Hydroxybenzylidene)amino]phenoxy}propyl)-1*H*-1,3-benzodiazol-2-yl]phenol

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Comment

In our ongoing studies on the synthesis, structural and spectroscopic characterization of the products derived from *N*¹-(3-(2-aminophenoxy)propyl)-benzene-1,2-diamine with aldehydes (Keypour *et al.*, 2009; Mahmoudi *et al.*, 2009) we report herein the crystal structure of the title compound, prepared by the reaction of *N*¹-(3-(2-aminophenoxy)propyl)-benzene-1,2-diamine with salicyl aldehyde.

The molecular structure of the title compound is shown in Fig. 1. The molecule adopts the *E* configuration with respect to the imine C=N bond. Two hydroxyl groups are located close to N atoms, and form strong intramolecular hydrogen bonds (Table 1).

Experimental

*N*¹-(3-(2-aminophenoxy)propyl)benzene-1,2-diamine (0.064 g, 0.25 mmol) in methanol (20 ml) was added dropwise with stirring to a solution of salicylaldehyde (0.061 g, 0.5 mmol) in methanol (30 ml). The mixture was refluxed for 12 h. Then, the solution volume was reduced to 10 ml by evaporation, and a precipitate was formed. This was filtered off, washed with ether, and dried *in vacuo*. Vapour diffusion of diethyl ether into a methanolic solution of the product afforded yellow crystals in 60% yield.

Refinement

All C-bonded H atoms positions were calculated and refined with a riding model and $U_{\text{iso}}(\text{H})$ parameters set to 1.2 times $U_{\text{eq}}(\text{carrier C atom})$. Hydroxyl H atoms also ride on their O atoms, with O—H bond lengths fixed to 0.84 Å and $U_{\text{iso}}(\text{H}) = 1.5 U_{\text{eq}}(\text{carrier O atom})$.

Figures

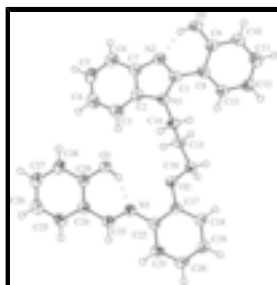


Fig. 1. A view of the structure of the title complex, with displacement ellipsoids drawn at the 50% probability level.

2-[1-(3-{2-[(2-Hydroxybenzylidene)amino]phenoxy}propyl)-1H-1,3-benzodiazol-2-yl]phenol

Crystal data

$C_{29}H_{25}N_3O_3$	$F(000) = 976$
$M_r = 463.52$	$D_x = 1.351 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
Hall symbol: -P 2ybc	Cell parameters from 16359 reflections
$a = 9.1097 (6) \text{ \AA}$	$\theta = 2.7\text{--}27.5^\circ$
$b = 18.1946 (11) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$c = 13.7769 (5) \text{ \AA}$	$T = 150 \text{ K}$
$\beta = 93.405 (4)^\circ$	Needle, yellow
$V = 2279.5 (2) \text{ \AA}^3$	$0.25 \times 0.12 \times 0.10 \text{ mm}$
$Z = 4$	

Data collection

Nonius KappaCCD diffractometer	5135 independent reflections
Radiation source: fine-focus sealed tube graphite	3083 reflections with $I > 2\sigma(I)$
Detector resolution: 9 pixels mm^{-1}	$R_{\text{int}} = 0.054$
φ scans and ω scans with κ offsets	$\theta_{\text{max}} = 27.5^\circ$, $\theta_{\text{min}} = 2.7^\circ$
Absorption correction: multi-scan (SORTAV; Blessing, 1995)	$h = -11 \rightarrow 11$
$T_{\text{min}} = 0.866$, $T_{\text{max}} = 0.993$	$k = -22 \rightarrow 23$
16359 measured reflections	$l = -17 \rightarrow 17$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.055$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.137$	H-atom parameters constrained
$S = 1.06$	$w = 1/[\sigma^2(F_o^2) + (0.0536P)^2 + 0.4485P]$
5135 reflections	where $P = (F_o^2 + 2F_c^2)/3$
318 parameters	$(\Delta/\sigma)_{\text{max}} = 0.001$
1 restraint	$\Delta\rho_{\text{max}} = 0.23 \text{ e \AA}^{-3}$
0 constraints	$\Delta\rho_{\text{min}} = -0.20 \text{ e \AA}^{-3}$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
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O1	0.98262 (18)	0.22668 (8)	0.88875 (9)	0.0465 (4)
H1O	0.9212	0.1922	0.8864	0.070*
O2	0.80830 (16)	0.02045 (7)	0.35975 (9)	0.0369 (4)
O3	0.69646 (17)	-0.13681 (8)	0.49252 (9)	0.0418 (4)
H2O	0.7212	-0.1156	0.4418	0.063*
N1	0.82575 (18)	0.08636 (9)	0.66217 (11)	0.0352 (4)
N2	0.80517 (19)	0.13040 (9)	0.81279 (11)	0.0378 (4)
N3	0.70199 (18)	-0.10938 (9)	0.31140 (11)	0.0323 (4)
C1	0.8674 (2)	0.13987 (11)	0.72906 (13)	0.0348 (5)
C2	0.7258 (2)	0.04118 (11)	0.70695 (14)	0.0355 (5)
C3	0.6454 (2)	-0.01894 (12)	0.67220 (16)	0.0424 (5)
H3A	0.6527	-0.0369	0.6079	0.051*
C4	0.5538 (2)	-0.05139 (12)	0.73642 (17)	0.0461 (6)
H4A	0.4961	-0.0926	0.7155	0.055*
C5	0.5439 (2)	-0.02520 (12)	0.83139 (17)	0.0462 (6)
H5A	0.4805	-0.0492	0.8736	0.055*
C6	0.6241 (2)	0.03446 (12)	0.86451 (16)	0.0428 (5)
H6A	0.6173	0.0520	0.9290	0.051*
C7	0.7156 (2)	0.06849 (11)	0.80071 (14)	0.0371 (5)
C8	0.9683 (2)	0.20146 (11)	0.71522 (14)	0.0342 (5)
C9	1.0249 (2)	0.24102 (12)	0.79778 (14)	0.0358 (5)
C10	1.1272 (2)	0.29677 (12)	0.78825 (15)	0.0406 (5)
H10A	1.1682	0.3211	0.8445	0.049*
C11	1.1694 (2)	0.31703 (12)	0.69784 (16)	0.0433 (5)
H11A	1.2402	0.3549	0.6920	0.052*
C12	1.1092 (2)	0.28247 (12)	0.61553 (15)	0.0411 (5)
H12A	1.1353	0.2979	0.5530	0.049*
C13	1.0116 (2)	0.22574 (11)	0.62434 (14)	0.0375 (5)
H13A	0.9720	0.2021	0.5671	0.045*
C14	0.8752 (2)	0.07122 (11)	0.56509 (13)	0.0354 (5)
H14A	0.8771	0.0174	0.5545	0.042*
H14B	0.9765	0.0900	0.5607	0.042*
C15	0.7747 (2)	0.10687 (12)	0.48571 (13)	0.0372 (5)
H15A	0.6749	0.0856	0.4877	0.045*
H15B	0.7675	0.1602	0.4991	0.045*
C16	0.8287 (2)	0.09626 (11)	0.38527 (14)	0.0375 (5)
H16A	0.7725	0.1280	0.3380	0.045*
H16B	0.9341	0.1095	0.3847	0.045*
C17	0.8190 (2)	0.00152 (11)	0.26392 (13)	0.0319 (5)
C18	0.8766 (2)	0.04669 (12)	0.19473 (14)	0.0382 (5)
H18A	0.9154	0.0936	0.2127	0.046*
C19	0.8774 (2)	0.02298 (12)	0.09868 (15)	0.0420 (5)
H19A	0.9156	0.0542	0.0510	0.050*
C20	0.8233 (2)	-0.04524 (12)	0.07220 (14)	0.0408 (5)
H20A	0.8249	-0.0611	0.0066	0.049*
C21	0.7667 (2)	-0.09051 (12)	0.14125 (14)	0.0367 (5)
H21A	0.7311	-0.1379	0.1230	0.044*
C22	0.7614 (2)	-0.06751 (11)	0.23715 (13)	0.0320 (5)
C23	0.5987 (2)	-0.15625 (11)	0.29380 (14)	0.0344 (5)

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H23A	0.5604	-0.1634	0.2288	0.041*
C24	0.5396 (2)	-0.19841 (11)	0.37172 (14)	0.0339 (5)
C25	0.4292 (2)	-0.25061 (12)	0.35230 (16)	0.0403 (5)
H25A	0.3930	-0.2584	0.2870	0.048*
C26	0.3717 (2)	-0.29106 (12)	0.42569 (17)	0.0458 (6)
H26A	0.2965	-0.3263	0.4114	0.055*
C27	0.4258 (2)	-0.27940 (12)	0.52143 (16)	0.0444 (6)
H27A	0.3875	-0.3073	0.5725	0.053*
C28	0.5338 (2)	-0.22813 (12)	0.54275 (15)	0.0407 (5)
H28A	0.5687	-0.2206	0.6083	0.049*
C29	0.5923 (2)	-0.18717 (11)	0.46906 (14)	0.0340 (5)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0556 (11)	0.0529 (11)	0.0310 (8)	-0.0016 (8)	0.0015 (7)	-0.0072 (7)
O2	0.0491 (9)	0.0346 (8)	0.0271 (7)	-0.0053 (7)	0.0029 (6)	-0.0018 (6)
O3	0.0502 (10)	0.0431 (9)	0.0320 (8)	-0.0091 (7)	0.0014 (6)	0.0014 (6)
N1	0.0373 (10)	0.0394 (10)	0.0289 (9)	0.0047 (8)	0.0017 (7)	-0.0045 (7)
N2	0.0435 (11)	0.0402 (11)	0.0298 (9)	0.0039 (9)	0.0032 (7)	-0.0015 (7)
N3	0.0351 (10)	0.0305 (10)	0.0314 (9)	0.0023 (8)	0.0033 (7)	0.0000 (7)
C1	0.0339 (12)	0.0397 (13)	0.0307 (11)	0.0079 (9)	-0.0001 (8)	-0.0026 (9)
C2	0.0312 (12)	0.0353 (12)	0.0401 (12)	0.0057 (9)	0.0023 (9)	0.0017 (9)
C3	0.0427 (13)	0.0416 (14)	0.0423 (13)	0.0072 (11)	-0.0021 (10)	-0.0059 (10)
C4	0.0393 (14)	0.0382 (13)	0.0604 (15)	0.0013 (11)	-0.0010 (11)	-0.0007 (11)
C5	0.0420 (14)	0.0399 (14)	0.0577 (15)	0.0057 (11)	0.0120 (11)	0.0111 (11)
C6	0.0461 (14)	0.0430 (14)	0.0398 (12)	0.0082 (11)	0.0076 (10)	0.0045 (10)
C7	0.0366 (12)	0.0392 (13)	0.0354 (12)	0.0060 (10)	0.0017 (9)	-0.0001 (9)
C8	0.0298 (11)	0.0366 (12)	0.0358 (11)	0.0035 (9)	-0.0005 (8)	-0.0022 (9)
C9	0.0358 (12)	0.0432 (13)	0.0283 (11)	0.0114 (10)	0.0000 (8)	-0.0022 (9)
C10	0.0368 (13)	0.0430 (13)	0.0410 (13)	0.0051 (11)	-0.0046 (9)	-0.0104 (10)
C11	0.0356 (13)	0.0408 (13)	0.0533 (14)	0.0008 (10)	0.0021 (10)	-0.0037 (10)
C12	0.0413 (13)	0.0444 (14)	0.0380 (12)	0.0029 (11)	0.0062 (9)	-0.0009 (10)
C13	0.0387 (13)	0.0415 (13)	0.0321 (11)	0.0024 (10)	0.0001 (9)	-0.0048 (9)
C14	0.0390 (12)	0.0399 (12)	0.0275 (11)	0.0067 (10)	0.0041 (8)	-0.0067 (9)
C15	0.0392 (12)	0.0392 (13)	0.0327 (11)	0.0039 (10)	-0.0007 (9)	-0.0045 (9)
C16	0.0437 (13)	0.0349 (12)	0.0335 (11)	-0.0050 (10)	-0.0003 (9)	-0.0007 (9)
C17	0.0323 (11)	0.0390 (12)	0.0244 (10)	0.0016 (9)	0.0011 (8)	-0.0013 (8)
C18	0.0417 (13)	0.0408 (13)	0.0325 (12)	-0.0043 (10)	0.0051 (9)	0.0012 (9)
C19	0.0448 (14)	0.0487 (14)	0.0334 (12)	-0.0011 (11)	0.0102 (9)	0.0040 (10)
C20	0.0458 (14)	0.0488 (14)	0.0286 (11)	0.0046 (11)	0.0080 (9)	-0.0028 (9)
C21	0.0398 (13)	0.0366 (13)	0.0336 (11)	0.0047 (10)	0.0024 (9)	-0.0041 (9)
C22	0.0308 (11)	0.0342 (12)	0.0310 (11)	0.0050 (9)	0.0027 (8)	0.0025 (8)
C23	0.0376 (12)	0.0333 (12)	0.0322 (11)	0.0058 (10)	0.0019 (9)	-0.0013 (9)
C24	0.0346 (12)	0.0297 (11)	0.0377 (12)	0.0050 (9)	0.0040 (9)	-0.0009 (9)
C25	0.0384 (13)	0.0353 (12)	0.0471 (13)	0.0004 (10)	0.0025 (10)	-0.0056 (10)
C26	0.0382 (14)	0.0345 (13)	0.0649 (16)	-0.0034 (10)	0.0058 (11)	0.0011 (11)
C27	0.0413 (14)	0.0381 (13)	0.0549 (14)	0.0044 (11)	0.0116 (10)	0.0133 (10)

C28	0.0406 (13)	0.0407 (13)	0.0414 (12)	0.0051 (11)	0.0061 (9)	0.0085 (10)
C29	0.0330 (12)	0.0295 (12)	0.0397 (12)	0.0026 (9)	0.0043 (9)	0.0005 (9)

Geometric parameters (Å, °)

O1—C9	1.358 (2)	C12—H12A	0.9500
O1—H10	0.8400	C13—H13A	0.9500
O2—C17	1.373 (2)	C14—C15	1.528 (3)
O2—C16	1.433 (2)	C14—H14A	0.9900
O3—C29	1.344 (2)	C14—H14B	0.9900
O3—H20	0.8400	C15—C16	1.508 (3)
N1—C1	1.378 (2)	C15—H15A	0.9900
N1—C2	1.397 (3)	C15—H15B	0.9900
N1—C14	1.462 (2)	C16—H16A	0.9900
N2—C1	1.326 (2)	C16—H16B	0.9900
N2—C7	1.395 (3)	C17—C18	1.385 (3)
N3—C23	1.282 (2)	C17—C22	1.402 (3)
N3—C22	1.409 (2)	C18—C19	1.392 (3)
C1—C8	1.469 (3)	C18—H18A	0.9500
C2—C3	1.386 (3)	C19—C20	1.377 (3)
C2—C7	1.392 (3)	C19—H19A	0.9500
C3—C4	1.384 (3)	C20—C21	1.381 (3)
C3—H3A	0.9500	C20—H20A	0.9500
C4—C5	1.400 (3)	C21—C22	1.390 (3)
C4—H4A	0.9500	C21—H21A	0.9500
C5—C6	1.371 (3)	C23—C24	1.449 (3)
C5—H5A	0.9500	C23—H23A	0.9500
C6—C7	1.392 (3)	C24—C25	1.397 (3)
C6—H6A	0.9500	C24—C29	1.412 (3)
C8—C13	1.406 (3)	C25—C26	1.379 (3)
C8—C9	1.417 (3)	C25—H25A	0.9500
C9—C10	1.388 (3)	C26—C27	1.397 (3)
C10—C11	1.376 (3)	C26—H26A	0.9500
C10—H10A	0.9500	C27—C28	1.374 (3)
C11—C12	1.381 (3)	C27—H27A	0.9500
C11—H11A	0.9500	C28—C29	1.391 (3)
C12—C13	1.372 (3)	C28—H28A	0.9500
C9—O1—H10	109.5	H14A—C14—H14B	107.9
C17—O2—C16	117.50 (14)	C16—C15—C14	112.85 (17)
C29—O3—H20	109.5	C16—C15—H15A	109.0
C1—N1—C2	106.32 (16)	C14—C15—H15A	109.0
C1—N1—C14	131.09 (17)	C16—C15—H15B	109.0
C2—N1—C14	122.48 (16)	C14—C15—H15B	109.0
C1—N2—C7	106.17 (16)	H15A—C15—H15B	107.8
C23—N3—C22	122.11 (16)	O2—C16—C15	107.68 (16)
N2—C1—N1	112.02 (18)	O2—C16—H16A	110.2
N2—C1—C8	121.00 (17)	C15—C16—H16A	110.2
N1—C1—C8	126.98 (18)	O2—C16—H16B	110.2
C3—C2—C7	122.6 (2)	C15—C16—H16B	110.2

supplementary materials

C3—C2—N1	131.05 (19)	H16A—C16—H16B	108.5
C7—C2—N1	106.32 (18)	O2—C17—C18	124.33 (18)
C4—C3—C2	116.2 (2)	O2—C17—C22	115.51 (17)
C4—C3—H3A	121.9	C18—C17—C22	120.12 (17)
C2—C3—H3A	121.9	C17—C18—C19	119.6 (2)
C3—C4—C5	121.8 (2)	C17—C18—H18A	120.2
C3—C4—H4A	119.1	C19—C18—H18A	120.2
C5—C4—H4A	119.1	C20—C19—C18	120.6 (2)
C6—C5—C4	121.2 (2)	C20—C19—H19A	119.7
C6—C5—H5A	119.4	C18—C19—H19A	119.7
C4—C5—H5A	119.4	C19—C20—C21	119.83 (19)
C5—C6—C7	118.0 (2)	C19—C20—H20A	120.1
C5—C6—H6A	121.0	C21—C20—H20A	120.1
C7—C6—H6A	121.0	C20—C21—C22	120.7 (2)
C6—C7—C2	120.2 (2)	C20—C21—H21A	119.6
C6—C7—N2	130.68 (19)	C22—C21—H21A	119.6
C2—C7—N2	109.13 (17)	C21—C22—C17	119.08 (18)
C13—C8—C9	116.56 (19)	C21—C22—N3	124.35 (18)
C13—C8—C1	124.48 (18)	C17—C22—N3	116.57 (16)
C9—C8—C1	118.95 (18)	N3—C23—C24	120.85 (18)
O1—C9—C10	117.15 (18)	N3—C23—H23A	119.6
O1—C9—C8	122.22 (19)	C24—C23—H23A	119.6
C10—C9—C8	120.63 (18)	C25—C24—C29	118.71 (18)
C11—C10—C9	120.45 (19)	C25—C24—C23	120.85 (18)
C11—C10—H10A	119.8	C29—C24—C23	120.43 (18)
C9—C10—H10A	119.8	C26—C25—C24	121.5 (2)
C10—C11—C12	120.2 (2)	C26—C25—H25A	119.2
C10—C11—H11A	119.9	C24—C25—H25A	119.2
C12—C11—H11A	119.9	C25—C26—C27	118.9 (2)
C13—C12—C11	119.8 (2)	C25—C26—H26A	120.6
C13—C12—H12A	120.1	C27—C26—H26A	120.6
C11—C12—H12A	120.1	C28—C27—C26	120.9 (2)
C12—C13—C8	122.18 (18)	C28—C27—H27A	119.6
C12—C13—H13A	118.9	C26—C27—H27A	119.6
C8—C13—H13A	118.9	C27—C28—C29	120.5 (2)
N1—C14—C15	111.78 (16)	C27—C28—H28A	119.7
N1—C14—H14A	109.3	C29—C28—H28A	119.7
C15—C14—H14A	109.3	O3—C29—C28	119.03 (18)
N1—C14—H14B	109.3	O3—C29—C24	121.48 (18)
C15—C14—H14B	109.3	C28—C29—C24	119.48 (19)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1—H10 \cdots N2	0.84	1.81	2.564 (2)	148
O3—H20 \cdots N3	0.84	1.80	2.548 (2)	148

Fig. 1

